Problem Set 1 The free electron gas

1 Particle in a box

Let us consider a free, nonrelativistic quantum particle with mass m and spin s confined in a cubic box with sides of length L.

- (a) Calculate the particle' wavefunctions and corresponding quantized energies for the cases of(i) hard wall and (ii) periodic boundary conditions.
- (b) Give (without any justification) the spin degeneracy factor η_s of each orbital eigenstate. How much is η_s for electrons, that are spin s = 1/2 elementary particles?
- (c) Considering from now on electrons, and using periodic boundary conditions, estimate the number $\mathcal{N}(E)$ of quantum states in the box that have an energy lower than E, in the regime where the box is sufficiently large such that $L \gg 2\pi\hbar/\sqrt{2mE}$. Show that in this limit, $\mathcal{N}(E)$ is proportional to the volume $V = L^3$ of the box, and that it does not depend on the choice of the boundary conditions.
- (d) Use the result of question (c) to calculate the density of states per unit volume g(E) (*i.e.*, the number of quantum states per unit energy and per unit volume).
- (e) Determine the density of states per unit surface and length, respectively, for free electrons in two and one dimensions.

2 Three-dimensional electron gas

- (a) Use the results of Exercise 1 to express the Fermi energy $E_{\rm F}$, the Fermi wavevector $k_{\rm F}$, and the Fermi velocity $v_{\rm F}$ of a three-dimensional electron gas as a function of the electron density n.
- (b) Show that the total kinetic energy of a three-dimensional electron gas containing N free electrons at zero temperature is given by $E_{\rm kin} = \frac{3}{5}NE_{\rm F}$.
- (c) The spatial density of atoms in copper is $8.45 \times 10^{28} \,\mathrm{m}^{-3}$. Estimate the values of the quantities in question (a) for the conduction band of copper, assuming a free electron gas and one conduction band electron per atom. Use the bare electron mass $m_{\rm e} = 9.11 \times 10^{-31} \,\mathrm{kg}$, and Planck's constant $\hbar = 1.05 \times 10^{-34} \,\mathrm{Js}$ (1 eV = $1.60 \times 10^{-19} \,\mathrm{J}$).