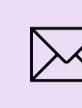


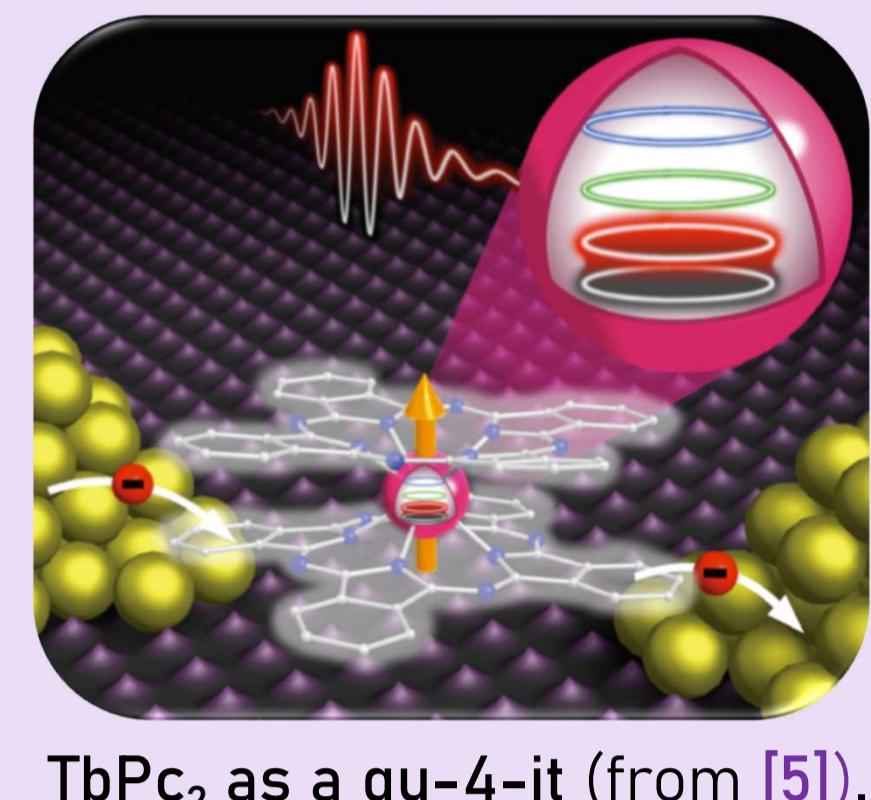
# Computation of the hyperfine structure of lanthanide-organic complexes used for Quantum Information Processing

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## Motivation



TbPc<sub>2</sub> as a qu-4-it (from [5]).

- Lanthanide-organic complexes ⇒ Quantum Information Processing.<sup>[1]</sup>
- High nuclear spin ⇒ high number of states.
- Ln<sup>3+</sup> in Ln-based complexes = 4f valence electrons ⇒ interesting optical and magnetic properties.<sup>[2]</sup>
- Nuclear properties + electronic properties ⇒ individually addressable hyperfine levels = qudits.<sup>[3,4]</sup>
- Complexes with Ln<sup>2+</sup> = additional 6s valence electron ⇒ penetration inside the nucleus ⇒ even larger hyperfine spacings.
- Playground to study the effect the internal structure of the nucleus has on electronic levels.

## Quantum Information

#	bits	qubits	qudits ( $d = 4$ )
1	0 ou 1	$\alpha_1 0\rangle + \alpha_2 1\rangle$	$\alpha_1 0\rangle + \alpha_2 1\rangle + \alpha_3 2\rangle + \alpha_4 3\rangle$
2	00, 01, 10 or 11	$\alpha_1 00\rangle + \alpha_2 01\rangle + \alpha_3 10\rangle + \alpha_4 11\rangle$	$\alpha_1 0000\rangle + \alpha_2 0001\rangle + \dots + \alpha_{16} 1111\rangle$
:	:	:	:
$N$	1 of $2^N$ combinaisons	$2^N$ probability amplitudes	$d^N$ probability amplitudes

- $d = 4$  : Grover's algorithm in only one molecule<sup>[4]</sup>.

+ Also useful for Q.E.C.\* [6]

Table 1: Comparison of the information density between bases of information.

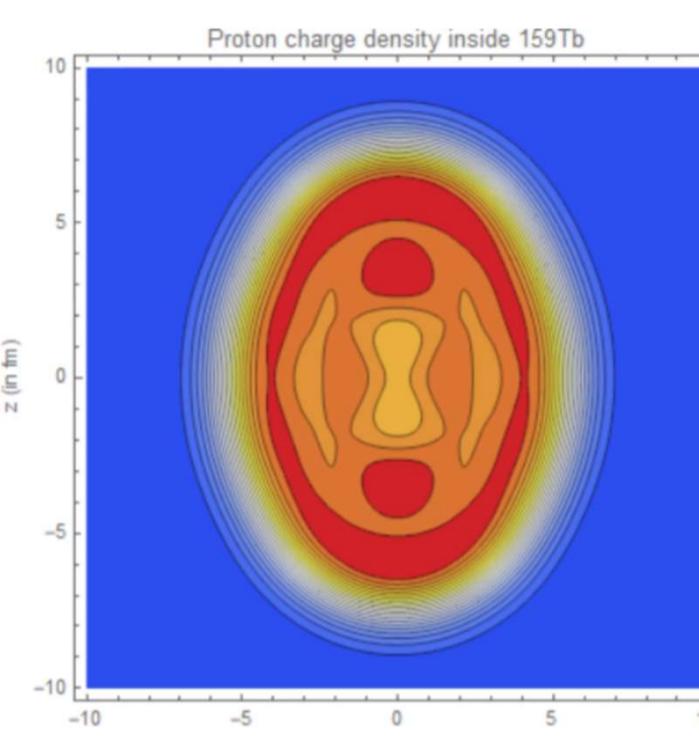
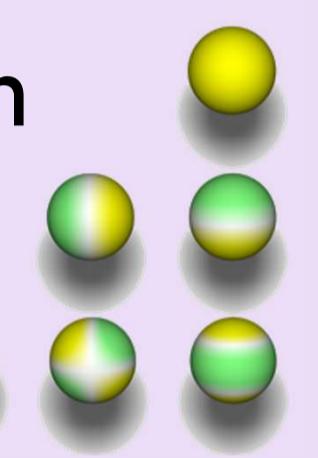


Figure 1: Skyrme-Hartree-Fock calculations : charge density inside <sup>159</sup>Tb.

## Spherical Harmonics Expansion

$$\frac{1}{|\mathbf{r}_e - \mathbf{r}_N|} = \sum_{k=0}^{\infty} \frac{r_e^k}{r_N^{k+1}} C_q^{(k)}(\theta_1, \phi_1) C_q^{(k)}(\theta_2, \phi_2)$$



## Nuclear aspect

### Electric Potential

$$Q_{in,q}^{(k)}(r) := \int_{R=0}^r \rho(\mathbf{R}) R^k C_q^{(k)}(\Theta, \Phi) d^3R$$

$$Q_{ex,q}^{(k)}(r) := \int_{R=r}^{+\infty} \frac{\rho(\mathbf{R})}{R^{k+1}} C_q^{(k)}(\Theta, \Phi) d^3R$$

$$\phi(\mathbf{r}) = \left( \frac{1}{r} Q_{in,0}^{(0)}(r) + Q_{ex,0}^{(0)}(r) \right) + C_0^{(2)}(\theta, \phi) \left( \frac{1}{r^3} Q_{in,0}^{(2)}(r) + r^2 Q_{ex,0}^{(2)}(r) \right) + \dots$$

### Mag. Vector Potential

$$\mathbf{A}(\mathbf{r}) = \frac{\mu_0}{4\pi} \int \frac{\mathbf{j}(\mathbf{R})}{|\mathbf{r} - \mathbf{R}|} d^3R$$

$$\mathbf{A}(\mathbf{r}) = \frac{1}{c^2} \frac{\mathbf{M}_{in}^{(1)}(r) \times \mathbf{r}}{r^3} + \frac{1}{c^2} (\mathbf{M}_{ex}^{(1)}(r) \times \mathbf{r}) + \dots$$

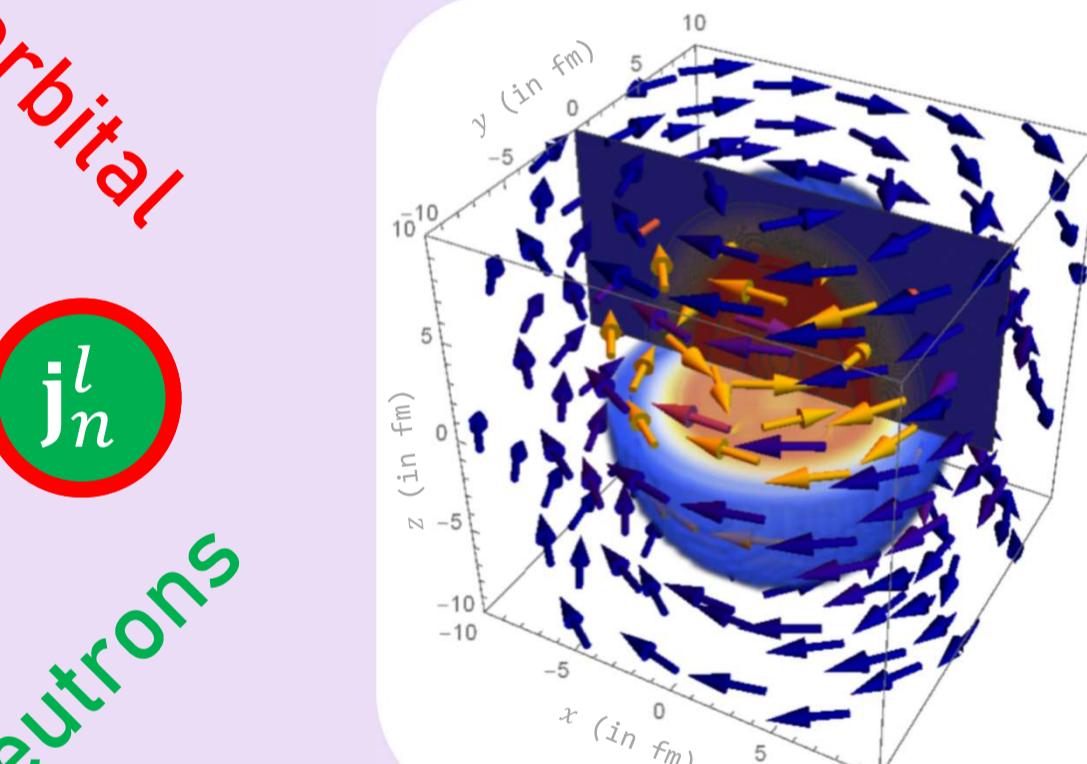


Figure 2: Different contributions to the total mag. vector pot. of the nucleus.

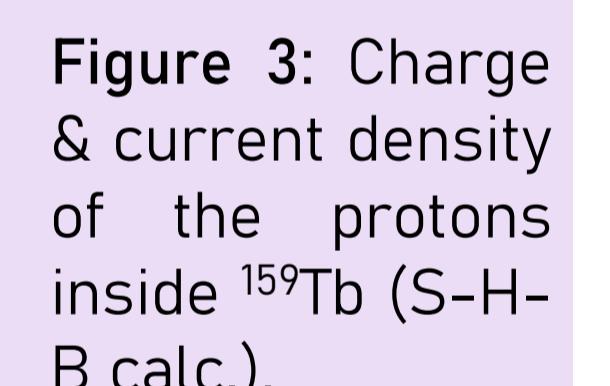


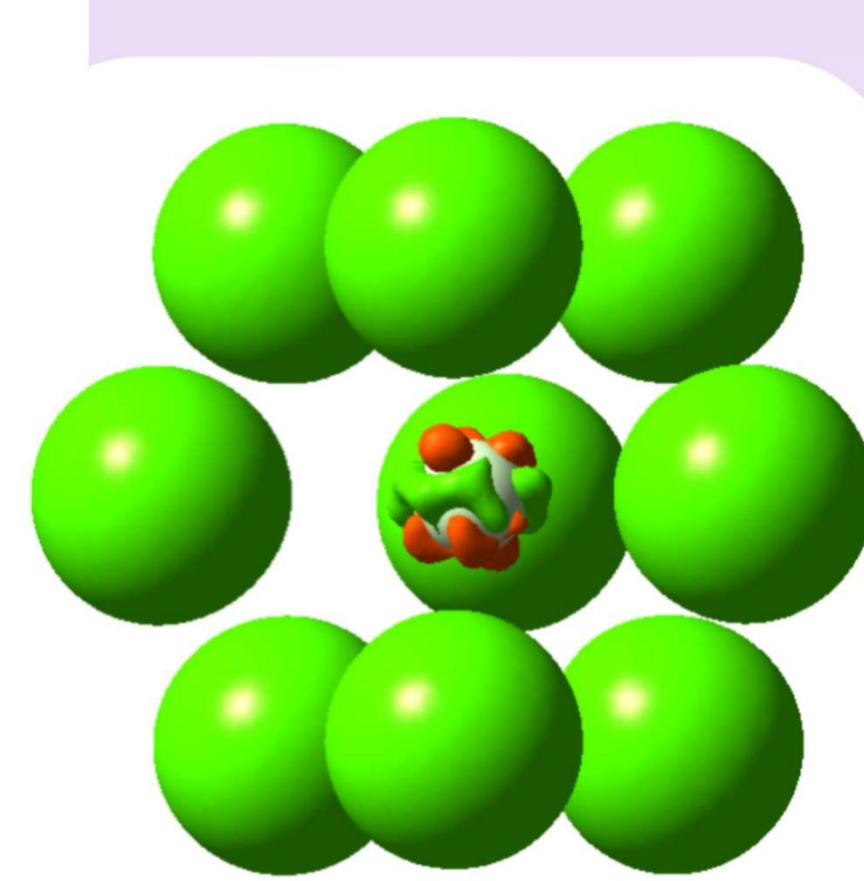
Figure 3: Charge & current density of the protons inside <sup>159</sup>Tb (S-H-B calc.).

## The crystal/molecule

### Ligand-Field Hamiltonian

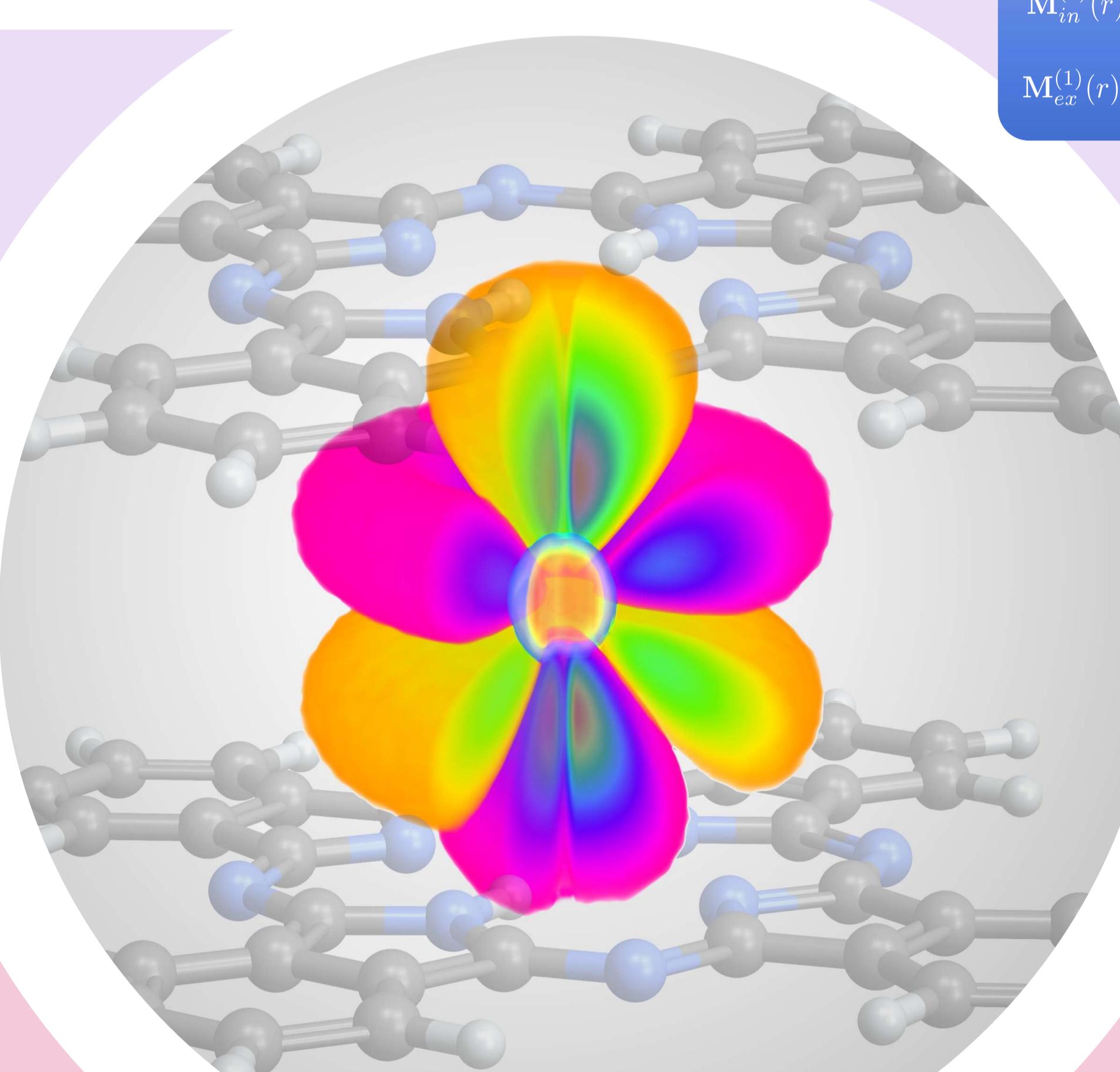
$$H_{LF} = \sum_k \sum_{q=-k}^k B_q^k C_q^k$$

= Encodes the symmetry + structure of the crystal /molecule

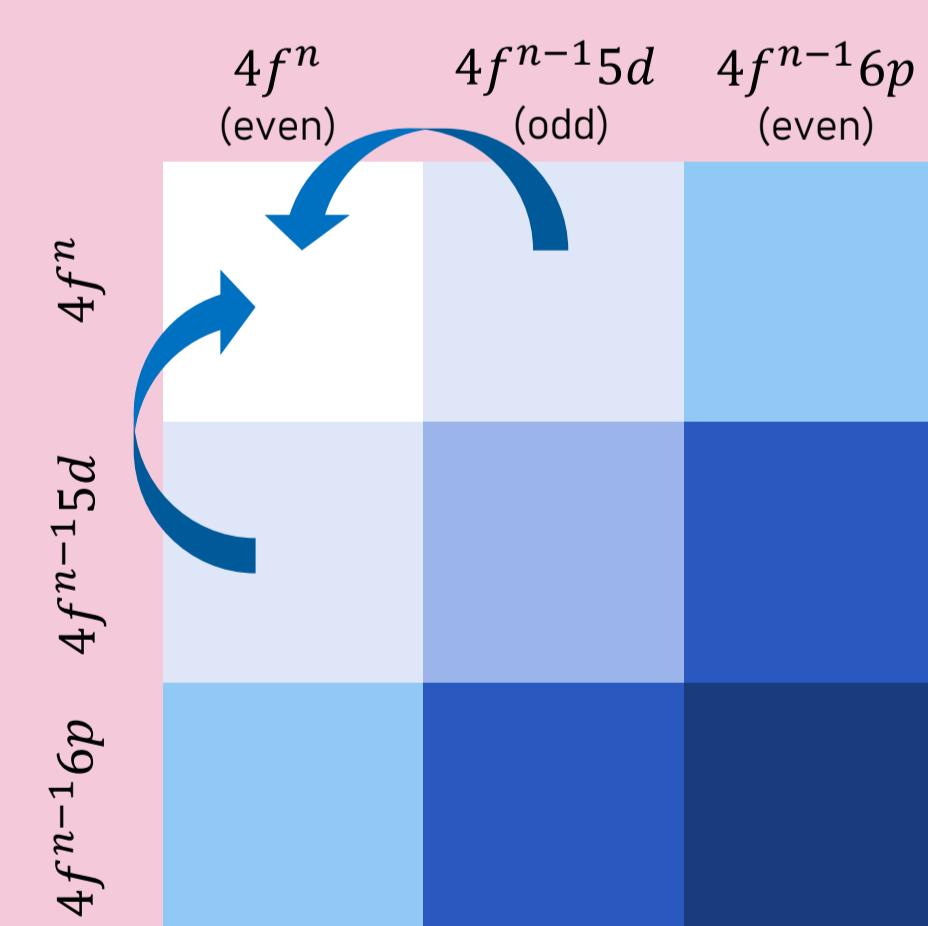


$$\int_0^{+\infty} \frac{\rho_L(\mathbf{R})}{R^{k+1}} C_q^{(k)}(\Theta, \Phi) d^3R$$

Figure 4: Electronic densities of the 9 Cl<sup>-</sup> ions surrounding the central Pr<sup>3+</sup> ion inside the PrCl<sub>3</sub> crystal (via DFT).



### Configuration Mixing



- Electrical control of the nuclear spin.<sup>[5]</sup>
- Optical f – f transitions.<sup>[2]</sup>
- Optical control of the nuclear spin ?

Non symmetric molecule  
⇒ States of undefined spatial parity.

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## The electrons – Atomistic approach

$j(\mathbf{R})$	$\rho(\mathbf{R})$	Point Charge	Charge distrib. model
Point Dipole		1	$(1+\epsilon_{BR}^{\text{model}})$
Current/Magnetiz. distribution Model		N/A	$(1+\epsilon_{BR}^{\text{model}})(1+\epsilon_{BW}^{\text{model}})$

Table 2: Correction factors to the hyperfine dipole constant  $A_{dip}$  from point-like to realistic models.

## Future Prospects

- Pursue the computation of ligand field parameters via multipole expansion.
- $A_{dip}$  &  $B_{quad}$  only in H-like free ions (single electron) for now...
  - Compute them with multi-electronic ions/atoms.
  - Go past the free ion Hamiltonian and add the ligand field.
- Compute f – f transition intensities.
- Take into account the excitations of the ligand.
- Compute the coherence times / oscillator strengths.
- Help experimental team... to implement a physical quantum error correcting code or other algorithms with qudits.

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